	Reception	Year 1	Year 2	Year 3	Year 4	Year 5	Year 6	Year 7	Year 8	Year 9	Year 10	Year 11
	Explore the	Plants:	Plants:	Plants:	Animals	Animals Including	Animals Including	Cells	Food and nutrition	Key concepts in	Health and disease	Exchange and transport in
Biology	natural world	Identify and	Observe and	Identify and	Including	Humans:	Humans:	Identify features of		Biology	Communicable and non-	animals
07	around them,	name a variety	describe how	describe the	Humans:	Describe the	Identify and name	animal and plant	What nutrients are and	Plant, animal and	communicable disease	Adaptations for transport
	making	of common wild	seeds and bulbs	functions of	Describe the	changes as	the main parts of	cells	what they do	bacterial cells		
	observations	and garden	grow into mature	different parts	simple functions	humans develop	the human			Specialised cells	Cardiovascular disease	The circulatory system
	and drawing	plants, including	plants	of flowering	of the basic parts	to old age	circulatory system,	Identify life	Components of a			
	pictures of	deciduous and	Find out and	plants: roots,	of the digestive		and describe the	processes	balanced diet	Using microscopes	Types of pathogens and	Structure of the heart
	animals and	evergreen trees	describe how	stem/trunk,	system in	Living Things and	functions of the	processes	bulunced diet		pathogenic diseases	Components of blood
	plants.		plants need	leaves and	humans	their Habitats:	heart, blood	Describe the organs	Steps in digestion and	Enzymes		components of blood
		Identify and	water, light and a	flowers	nununs	Describe the	vessels and blood	in some organ	the organs involved	Testing foods	Virus life cycles	Aerobic and anaerobic
	To be able to	describe the	suitable	nowers	Identify the	differences in the	Recognise the	systems	(and the role of	resting joous	Virus inje cycles	respiration
	name some	basic structure		Explore the	different types	life cycles of a	impact of diet,	Systems		Transporting	Defences against disease	
	objects found	of a variety of	temperature to		of teeth in			Sovual	enzymes)	Transporting substances (diffusion,	Defences against disease	Ecosystems and material cycles
	in the natural	common	grow and stay	requirements of		mammal, an	exercise, drugs	<u>Sexual</u>	Absorption		Change in marking marking	What an ecosystem is
	world e.g.	flowering plants,	healthy.	plants for life	humans and	amphibian, an	and lifestyle on the	reproduction in	Absorption	osmosis and active	Steps in making medicine	
	conker,	including trees.		and growth (air,	their simple	insect and a bird	way their bodies	animals	(adaptations of the	transport)		Energy transfer
	acorns, pine		Animals	light, water,	functions		function	Steps in animals	intestine)		Antibiotics and	
	cone,	Animals	Including	nutrients from		Describe the life	Describe the ways	sexual reproduction		Cells and control	monoclonal antibodies	Abiotic factors
	chestnut	Including	Humans:	soil, and room to	Construct and	process of	in which nutrients	(humans)	Plants and their	Stages in mitosis		
	Talvaavvukat	Humans:	Notice that	grow) and how	interpret a	reproduction in	and water are		<u>reproduction</u>		Plant structures and	Biotic factors
	To know what	Identify and	animals,	they vary from	variety of food	some plants and	transported within	Name the male and		Growth in animals and	functions	Accessing a literian
	a plant needs	name a variety	including	plant to plant	chains,	animals.	animals, including	female	Classification of plants	plants	Recall the photosynthesis	Assessing pollution
	to grow and to be able to	of common	humans, have		identifying		humans.	reproductive organs			equation	Parasitism and mutualism
	name some	animals	offspring which	Investigate the	producers,				What biodiversity	Stem cells		
	plants	including fish, amphibians,	grow into adults	way in which	predators and		Living Things and	Describe stages in	means and why it is		Factors affecting	Biodiversity threats and
	piants	reptiles, birds	Find out about	water is	prey		<u>their Habitat</u>	pregnancy and birth	important	The nervous system	photosynthesis	conservation
	To be able to	and mammals	and describe the	transported			Describe how			including types of		
	categorise	and manimals	basic needs of	within plants	Living Things		living things are	Muscles and bones	Types of plant	nerves, reflex arc and	Understand how plants	Food security
	farm and wild	Identify and	animals,		and their		classified into		reproduction	synapse	absorb water and mineral	
	animals and	name a variety	including	Explore the part	Habitats:		broad groups	Describe how			ions	Material cycles (water,
	pets	of common	humans, for	that flowers play	Recognise that		according to	muscles are used	Steps in plant	Structure and function		nitrogen and carbon)
	F	animals that are	survival (water,	in the life cycle	living things can		common	for breathing	reproduction	of the brain	Transpiration and	
	To be able to	carnivores,	food and air)	of flowering	be grouped in a		observable				translocation	Rates of decomposition
	name and	herbivores and	Describe the	plants, including	variety of ways		characteristics and	Describe how	Breathing and	Structure and function		
	describe	omnivores	importance for	pollination, seed			based on	oxygen gets to	respiration	of the eye	Plant adaptations	
	some		humans of	formation and	Explore and use		similarities and	muscles				
	common bugs	Describe and	exercise, eating	seed dispersal.	classification		differences,		Describe aerobic	Genetics	Plant hormones	
		compare the	the right		keys to help		including	Structure and	respiration	Sexual and asexual		
		structure of a	amounts of	Animals	group, identify		microorganisms,	function of the		reproduction	Animal coordination and	
		variety of	different types of	Including	and name a		plants and animals	skeleton	Understand how gases		control	
		common	food, and	Humans:	variety of living		Give reasons for		are exchanged in the	Steps in meiosis	What hormones are and	
		animals (fish,	hygiene.	Identify that	things in their		classifying plants	Ecosystems	lungs		examples of human	
		amphibians,		animals,	local and wider		and animals based			Structure of DNA	hormones	
		reptiles, birds	Living Things and	including	environment		on specific	Types of variation	The role of blood in			
		and mammals,	their Habitats:	humans, need	citvitorinicité		characteristics.		carrying oxygen	Steps in DNA	Metabolic rate	
		including pets)	Explore and	the right types	Recognise that			Adaptations of	Carlying Oxygen	extraction		
		lala a tit	compare the	and amount of	environments		Evolution and	plants and animals	Describe anaerobic		Hormones and the	
		Identify, name,	differences	nutrition, and	can change and		Inheritance:		respiration	Protein synthesis	menstrual cycle	
		draw and label	between things	that they cannot	that this can		Recognise that	Daily and seasonal	respiration			
		the basic parts	-				living things have			Genetics (inheritance	Control of blood glucose	
		of the human	that are living,	make their own	sometimes pose			changes and	Unicellular organisms	-	Control of blood glucose	
		body and say	dead, and things	food; they get	dangers to living		changed over time	adaptations	Lindonaka mala 1976 - 1	and mutation)	and diabetes (types 1 and	
1		which part of the	that have never	nutrition from	things.		and that fossils	Transferrate	Understand different	Mandal	2)	
1		body is associated with	been alive	what they eat			provide	Transfer of energy	types of unicellular	Mendel and genetics		
		each sense.	Identify that				information about	in food chains	organisms		Thermoregulation	
1		Cach Sense.	most living things				living things that			Variation		

	1	1		1		1					1	
			live in habitats to	Identify that			inhabited the		Understand the		Osmoregulation and	
			which they are	humans and			Earth millions of		structure of	Natural selection and	structure and function of	
			suited and	some other			years ago		microscopic fungi	GM	the kidneys	
			describe how	animals have			Recognise that			Evidence for human		
			different habitats	skeletons and			living things		Describe the structure	evolution		
			provide for the	muscles for			produce offspring		of protoctists			
			basic needs of	support,			of the same kind,			Understand Darwin's		
			different kinds of	protection and			but normally		Understand how	theory of evolution		
			animals and	1.			offspring vary and		unicellular organisms			
				movement.						The electification		
			plants, and how				are not identical to		can be helpful and	The classification		
			they depend on				their parents		harmful	system		
			each other				Identify how					
			Identify and				animals and plants		The carbon cycle	The differences		
			name a variety of				are adapted to suit			between breeds and		
			plants and				their environment			varieties		
			animals in their				in different ways			Tissue culture		
			habitats,				and that					
			including				adaptation may			Genetic modification		
			microhabitats				lead to evolution.			including benefits and		
			Describe how							drawbacks		
			animals obtain									
			their food from							Fertilisers and		
			plants and other							biological control		
			animals, using							Siciogrea control		
			the idea of a									
			simple food									
			chain, and									
			identify and									
			name different									
			sources of food.									
	To be able to	Everyday	Everyday	Rocks:	States of	Properties and		Mixtures and	Combustion	States of matter;	Calculations involving	Groups in the Periodic table,
Chemistry	comment on	Materials:	<u>Everyday</u> <u>Materials:</u>	Compare and	States of Matter:	changes of		Mixtures and separation		separating mixtures	masses	rates of reaction
Chemistry	comment on the changes		Everyday						Combustion Describe what happens			
Chemistry	comment on the changes of the	Materials: Distinguish between an	Everyday Materials: Identify and compare the	Compare and	Matter:	changes of				separating mixtures	masses	rates of reaction Trends in groups 1,7 and 0
Chemistry	comment on the changes of the properties of	Materials: Distinguish	Everyday Materials: Identify and	Compare and group together	Matter: Compare and	<u>changes of</u> <u>Materials:</u>		separation	Describe what happens	separating mixtures	masses Calculating empirical	rates of reaction
Chemistry	comment on the changes of the properties of objects e.g	Materials: Distinguish between an	Everyday Materials: Identify and compare the	Compare and group together different kinds	Matter: Compare and group materials	changes of <u>Materials:</u> Compare and		separation	Describe what happens	separating mixtures States of matter	masses Calculating empirical	rates of reaction Trends in groups 1,7 and 0 Rates of reaction
Chemistry	comment on the changes of the properties of objects e.g paint, ice and	Materials: Distinguish between an object and the	Everyday Materials: Identify and compare the suitability of a	Compare and group together different kinds of rocks on the	Matter: Compare and group materials together, according to	changes of <u>Materials:</u> Compare and group together		separation Define mixtures	Describe what happens when fuels are burned	separating mixtures States of matter	masses Calculating empirical formulae	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of
Chemistry	comment on the changes of the properties of objects e.g	Materials: Distinguish between an object and the material from	Everyday Materials: Identify and compare the suitability of a variety of everyday	Compare and group together different kinds of rocks on the basis of their appearance and	Matter: Compare and group materials together, according to whether they are	changes of Materials: Compare and group together everyday materials on the		separation Define mixtures Describe solutions,	Describe what happens when fuels are burned Write word equations	separating mixtures States of matter Mixtures Separating mixtures	<u>masses</u> Calculating empirical formulae Law of conservation of	rates of reaction Trends in groups 1,7 and 0 Rates of reaction
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food	Materials: Distinguish between an object and the material from which it is made	Everyday Materials: Identify and compare the suitability of a variety of everyday materials,	Compare and group together different kinds of rocks on the basis of their appearance and simple physical	Matter: Compare and group materials together, according to whether they are solids, liquids or	changes ofMaterials:Compare andgroup togethereverydaymaterials on thebasis of their		separation Define mixtures Describe solutions, solutes and solvents	Describe what happens when fuels are burned Write word equations for oxidation	separating mixtures States of matter Mixtures Separating mixtures (chromatography,	masses Calculating empirical formulae Law of conservation of mass	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of reaction
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about	Materials: Distinguish between an object and the material from which it is made Identify and name	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood,	Compare and group together different kinds of rocks on the basis of their appearance and	Matter: Compare and group materials together, according to whether they are	changes of Materials: Compare and group together everyday materials on the basis of their properties,		separation Define mixtures Describe solutions, solutes and solvents Use evaporation to	Describe what happens when fuels are burned Write word equations for oxidation Understand how	separating mixtures States of matter Mixtures Separating mixtures (chromatography, filtration, distillation,	<u>masses</u> Calculating empirical formulae Law of conservation of	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic,	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties	Matter: Compare and group materials together, according to whether they are solids, liquids or gases	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their		separation Define mixtures Describe solutions, solutes and solvents	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to	separating mixtures States of matter Mixtures Separating mixtures (chromatography,	massesCalculating empiricalformulaeLaw of conservation ofmassConcentration calculations	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of reaction
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock,	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness,		separation Define mixtures Describe solutions, solutes and solvents Use evaporation to separate solutions	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global	separating mixtures States of matter Mixtures Separating mixtures (chromatography, filtration, distillation, crystallisation)	masses Calculating empirical formulae Law of conservation of mass	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of reaction Catalysts and activation energy
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g.	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials,	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility,		separation Define mixtures Describe solutions, solutes and solvents Use evaporation to separate solutions Use	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to	separating mixtures States of matter Mixtures Separating mixtures (chromatography, filtration, distillation,	massesCalculating empirical formulaeLaw of conservation of massConcentration calculationsMoles	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of reaction Catalysts and activation energy Exothermic and endothermic
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g. reflective,	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood,	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency,		separation Define mixtures Describe solutions, solutes and solvents Use evaporation to separate solutions Use chromatography to	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global warming	separating mixturesStates of matterMixturesSeparating mixtures(chromatography,filtration, distillation,crystallisation)Making drinking water	massesCalculating empirical formulaeLaw of conservation of massConcentration calculationsMolesElectrolysis and metals	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of reaction Catalysts and activation energy Exothermic and endothermic
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g.	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood, plastic, glass,	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for particular uses	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are formed when	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state when they are	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity		separation Define mixtures Describe solutions, solutes and solvents Use evaporation to separate solutions Use	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global	separating mixtures States of matter Mixtures Separating mixtures (chromatography, filtration, distillation, crystallisation)	massesCalculating empirical formulaeLaw of conservation of massConcentration calculationsMoles	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of reaction Catalysts and activation energy Exothermic and endothermic reactions
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g. reflective,	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood, plastic, glass, metal, water, and	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for particular uses Find out how the	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are formed when things that have	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state when they are heated or	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and		separation Define mixtures Describe solutions, solutes and solvents Use evaporation to separate solutions Use chromatography to separate colours	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global warming <u>The Periodic table</u>	separating mixturesStates of matterMixturesSeparating mixtures(chromatography, filtration, distillation, crystallisation)Making drinking waterAtomic structure	massesCalculating empiricalformulaeLaw of conservation ofmassConcentration calculationsMolesElectrolysis and metalsPrinciples of electrolysis	rates of reactionTrends in groups 1,7 and 0Rates of reactionFactors affecting rates of reactionCatalysts and activation energyExothermic and endothermic reactionsFuels, earth and atmosphere
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g. reflective,	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood, plastic, glass,	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for particular uses Find out how the shapes of solid	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are formed when things that have lived are	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state when they are heated or cooled, and	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and		separationDefine mixturesDescribe solutions, solutes and solventsUse evaporation to separate solutionsUse chromatography to separate coloursUse distillation to	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global warming The Periodic table Describe Dalton's	separating mixturesStates of matterMixturesSeparating mixtures(chromatography,filtration, distillation,crystallisation)Making drinking water	massesCalculating empirical formulaeLaw of conservation of massConcentration calculationsMolesElectrolysis and metals	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of reaction Catalysts and activation energy Exothermic and endothermic reactions Fuels, earth and atmosphere Hydrocarbons in crude oil and natural gas
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g. reflective,	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood, plastic, glass, metal, water, and rock	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for particular uses Find out how the shapes of solid objects made	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are formed when things that have lived are trapped within	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state when they are heated or cooled, and measure or	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and response to		separation Define mixtures Describe solutions, solutes and solvents Use evaporation to separate solutions Use chromatography to separate colours	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global warming <u>The Periodic table</u>	separating mixturesStates of matterMixturesSeparating mixtures(chromatography, filtration, distillation, crystallisation)Making drinking waterAtomic structureStructure of the atom	massesCalculating empiricalformulaeLaw of conservation ofmassConcentration calculationsMolesElectrolysis and metalsPrinciples of electrolysisProducts from electrolysis	rates of reactionTrends in groups 1,7 and 0Rates of reactionFactors affecting rates of reactionCatalysts and activation energyExothermic and endothermic reactionsFuels, earth and atmosphere Hydrocarbons in crude oil and natural gasFractional distillation of crude
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g. reflective,	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood, plastic, glass, metal, water, and rock Describe the	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for particular uses Find out how the shapes of solid objects made from some	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are formed when things that have lived are	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state when they are heated or cooled, and measure or research the	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and		separationDefine mixturesDescribe solutions, solutes and solventsUse evaporation to separate solutionsUse chromatography to separate coloursUse distillation to produce pure water	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global warming The Periodic table Describe Dalton's atomic model	separating mixturesStates of matterMixturesSeparating mixtures(chromatography, filtration, distillation, crystallisation)Making drinking waterAtomic structureStructure of the atomUsing the periodic	massesCalculating empiricalformulaeLaw of conservation ofmassConcentration calculationsMolesElectrolysis and metalsPrinciples of electrolysis	rates of reaction Trends in groups 1,7 and 0 Rates of reaction Factors affecting rates of reaction Catalysts and activation energy Exothermic and endothermic reactions Fuels, earth and atmosphere Hydrocarbons in crude oil and natural gas
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g. reflective,	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood, plastic, glass, metal, water, and rock Describe the simple physical	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for particular uses Find out how the shapes of solid objects made from some materials can be	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are formed when things that have lived are trapped within rock	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state when they are heated or cooled, and measure or research the temperature at	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and response to magnets		separationDefine mixturesDescribe solutions, solutes and solventsUse evaporation to separate solutionsUse chromatography to separate coloursUse distillation to	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global warming The Periodic table Describe Dalton's atomic model Describe the	separating mixturesStates of matterMixturesSeparating mixtures(chromatography, filtration, distillation, crystallisation)Making drinking waterAtomic structureStructure of the atomUsing the periodic table to see atomic	massesCalculating empiricalformulaeLaw of conservation ofmassConcentration calculationsMolesElectrolysis and metalsPrinciples of electrolysisProducts from electrolysisThe reactivity series	rates of reactionTrends in groups 1,7 and 0Rates of reactionFactors affecting rates of reactionCatalysts and activation energyExothermic and endothermic reactionsFuels, earth and atmosphere Hydrocarbons in crude oil and natural gasFractional distillation of crude oil
Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g. reflective,	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood, plastic, glass, metal, water, and rock Describe the simple physical properties of a	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for particular uses Find out how the shapes of solid objects made from some materials can be changed by	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are formed when things that have lived are trapped within rock Recognise that	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state when they are heated or cooled, and measure or research the	changes of Materials: Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and response to magnets Know that some		separation Define mixtures Describe solutions, solutes and solvents Use evaporation to separate solutions Use chromatography to separate colours Use distillation to produce pure water Acids and alkalis	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global warming <u>The Periodic table</u> Describe Dalton's atomic model Describe the differences between	separating mixturesStates of matterMixturesSeparating mixtures(chromatography, filtration, distillation, crystallisation)Making drinking waterAtomic structureStructure of the atomUsing the periodic	massesCalculating empiricalformulaeLaw of conservation ofmassConcentration calculationsMolesElectrolysis and metalsPrinciples of electrolysisProducts from electrolysis	rates of reactionTrends in groups 1,7 and 0Rates of reactionFactors affecting rates of reactionCatalysts and activation energyExothermic and endothermic reactionsFuels, earth and atmosphere Hydrocarbons in crude oil and natural gasFractional distillation of crude
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Chemistry	comment on the changes of the properties of objects e.g paint, ice and food Talk about some properties of materials e.g. reflective,	Materials: Distinguish between an object and the material from which it is made Identify and name a variety of everyday materials, including wood, plastic, glass, metal, water, and rock Describe the simple physical properties of a variety of everyday	Everyday Materials: Identify and compare the suitability of a variety of everyday materials, including wood, metal, plastic, glass, brick, rock, paper and cardboard for particular uses Find out how the shapes of solid objects made from some materials can be changed by squashing, bending, twisting	Compare and group together different kinds of rocks on the basis of their appearance and simple physical properties Describe in simple terms how fossils are formed when things that have lived are trapped within rock Recognise that soils are made from rocks and	Matter: Compare and group materials together, according to whether they are solids, liquids or gases Observe that some materials change state when they are heated or cooled, and measure or research the temperature at which this happens in degrees Celsius	changes of Materials:Compare and group together everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and response to magnetsKnow that some materials will dissolve in liquid to form a		separationDefine mixturesDescribe solutions, solutes and solventsUse evaporation to separate solutionsUse chromatography to separate coloursUse distillation to produce pure waterAcids and alkalis Know what some hazards are and	Describe what happens when fuels are burned Write word equations for oxidation Understand how combustion can lead to air pollution and global warming The Periodic table Describe Dalton's atomic model Describe the differences between chemical properties and physical trends	separating mixturesStates of matterMixturesSeparating mixtures(chromatography, filtration, distillation, crystallisation)Making drinking waterAtomic structureStructure of the atomUsing the periodic table to see atomic number and massIsotopes	massesCalculating empiricalformulaeLaw of conservation ofmassConcentration calculationsMolesElectrolysis and metalsPrinciples of electrolysisProducts from electrolysisThe reactivity seriesExtracting metals fromores	rates of reactionTrends in groups 1,7 and 0Rates of reactionFactors affecting rates of reactionCatalysts and activation energyExothermic and endothermic reactionsFuels, earth and atmosphere Hydrocarbons in crude oil and natural gasFractional distillation of crude oilBreaking down hydrocarbonsComplete and incomplete combustion

variety of everyday materials on the basis of their simple physical properties		evaporation and condensation in the water cycle and associate the rate of evaporation with	substance from a solution Use knowledge of solids, liquids and gases to decide		Use the pH scale to measure acidity and alkalinity	Metals and their uses Describe the properties of metals Describe the reactions	How elements are arranged in groups and periods Atomic number and the Periodic table	Dynamic equilibrium Transition metals Corrosion	Hydrocarbons, alcohols, carboxy polymers Structure and reactions of alkanes and alkenes
		temperature.	how mixtures might be separated, including through filtering, sieving and evaporating		The particle model Draw solids, liquids and gases using the particle model	of metals with water and acids Describe the differences between alloys and pure metals	Electron configuration Bonding Ionic bonding and properties of ionic compounds	Electroplating The process of alloying Uses of metals and their alloys	Ethanol production Alcohols Carboxylic acids Polymerisation (addition and condensation)
			Give reasons, based on evidence from comparative and fair tests, for the particular uses of everyday materials, including metals, wood and plastic Demonstrate that dissolving, mixing and changes of state are reversible changes Explain that some changes result in the formation of new materials, and that this kind of change is not usually reversible, including changes associated with burning and the action of acid on		Understand the properties of solids, liquids and gases Describe Brownian motion and the evidence for this Describe diffusion in fluids Understand how air pressure is caused. Atoms, elements and molecules What air is made of Where elements come from Metals and non- metals What compounds are What happens in	Rocks Uses of rocks e.g. in building How igneous, metamorphic and sedimentary rocks are formed The fossil record Getting materials from the Earth	Covalent bonding Properties of molecular compounds Allotropes of carbon Properties of metals Advantages and disadvantages of bonding models Acids and alkalis Definitions of acids, alkalis and indicators The pH scale Neutralisation (with bases, carbonates and metals) Solubility	Quantitative analysis, equilibria, chemical and fuel cells • Yields • Atom economy • Concentrations • Titration calculations • Molar volumes of gases • Fertilisers and the Haber process • Factors affecting equilibrium • Chemical cells and fuel cells	Polymer problems Qualitative analysis and materials Flame tests and photometry Tests for positive ions Tests for negative ions Choosing materials Composite materials Nanoparticles
Seasonal Changes:	Light: Recognise that	<u>Sound:</u> Identify how	soda. <u>Forces:</u> Explain that	Light: Recognise that	Energy Know that food	Fluids	Conservation of Energy	Light and the Electromagnetic Spectrum	Electricity and Circuits Electric circuits
Observe changes across the four seasons	they need light in order to see things and that dark is the	sounds are made, associating some of them with	unsupported objects fall towards the Earth because of the	light appears to travel in straight lines Use the idea that	contains energy and how this can be measured	Using the particle model Describe changes of	Energy Stores and Transfers Energy Efficiency	Ray diagrams Colour	Current and potential difference Current charge and energy
Observe and describe weather	absence of light Notice that light	something vibrating	force of gravity acting between the Earth and the	light travels in straight lines to explain that	Understand energy transfers and stores	state Describe how pressure	Keeping Warm	Lenses EM waves	Resistance
	everyday materials on the basis of their simple physical properties.	everyday       materials on the         basis of their       simple physical         properties.       Image: Changes in the image:	everyday       condensation in         materials on the basis of their simple physical properties.       condensation in the water cycle and associate the rate of evaporation with temperature.         properties.       light:       source         Seasonal       light:       Sound: light: New York of the water	everyday materials on the basis of their simple physical properties.solutionsolutionproperties.Image: SolutionUse knowledge of solids, liquids and gesto decide how mixtures might be separated, including through filtering, sieving and evaporating and that this kind of change is not usuapy reversible, including changes associated with burning and that excision of action of the with something and that of the with evant of the with proces of the action of action of the with something and that of the with evant of the with proces of the with something and that of the with evant of the with of the with evant of the with something and that of the with evant of the with of the with evant of the with proces of the solution of the with something and that of the with evant of the with of the with evant of the with evant of the with of the with evant of	everyday materials on the basis of their simple physical properties.condensation in the water cycle and associate the rate of evaporation with emperature.solutionUse knowledge of solution, gases to decide how mixtures might be separated, including through filtering, sieving and evaporation and evaporation based on evidence from comparative and fair tests, for the particular uses of everyday materials, including metals, wood and plasticUse knowledge of solution, gases to decide how mixtures might be separated, including through filtering, sieving and evaporatingSeasonal Changes: Observe changes across the four seasonsLight: Recognise that they need light in order to see things and that somethingSound: the form with sounds are made, associating some thing sancting they need to things and that they need light across the four seasonsLight: Recognise that they need light arcs and they need light in order to see things and that some sociate are sociated with burning and the association of acid on bicarbonate of sounds are made, association of acid on they need light association some of them with of them with objects fail towards the Earth light appears to tavel in traight light appears to tavel in traight light appears to tavel in the light appears to tavel in traight light appears to tavel in traight light appears to tavel in trai	everyday materials on the basis of their simple physical properties.Use the phi scale to and associate of evaporation with 	everydy materials on the' ismige physical symple physical properties.Lisher hasso of the' and associate properties.condensation in the water of and associate the rate of evaporation with separated, including through and evaporating.condensation in humeres asses to deide how mittres separated, including through and evaporating.Use the phase to the mesure satisfies and associate mesure satisfies and gases to deide how mittres including through particle modelDescribe the reactions of metarilistion metarilistic me	ercryda hasis of trier apoperties.karren of the water cycle and associate apoperties.solution the water cycle use knowledge of osiles, liquids of apose of access to decide how matures apoperties.Use the space of and association with the rate of exaportion with the rate of the rate of the rate of the rate of the reaction with the reaction with<	emerging microticits on the basis of the far and sociality properties.solutionsolutionsolutionproperties of metals properties of metals of metals with water of metals wit

To categorise	I	Pocognico that	through a	Identify the	out or rofloat light	Understand sources	Describe forces in	Non-renewable		Transferring energy by
objects into		Recognise that	through a	effects of air	out or reflect light	of renewable			Using long waves	• • •
some		light from the	medium to the		into the eye Explain that we		floating and sinking	resources	Radiation and	electricity
characteristics		sun can be	ear	resistance, water	•	energy	Drag as a turna of	Renewable resources		
e.g. sinking		dangerous and	Find patterns	resistance and friction, that act	see things because light travels from	Current electricity	Drag as a type of	Reflewable resources	temperature	Electrical safety
and floating		that there are	-	,	•	Current electricity	friction	Mation	Lising short wayas	Magnatism and the Motor F
3		ways to protect their eyes	between the	between moving surfaces	light sources to our eyes or from	Understand how to	Light	Motion Vectors and Scalars	Using short waves	Magnetism and the Motor E
		their eyes	pitch of a sound and features of	Surraces	-	make a series and	Light	Vectors and Scalars	EM dangers	Magnets and magnetic fields
		Pocognico that	the object that	Recognise that	light sources to objects and then	parallel circuit	Describe luminous and	Distance time graphs	Elvi ualigers	Magnets and magnetic neids
		Recognise that shadows are	produced it	some	to our eyes	parallel circuit	non-luminous objects	Distance time graphs	Radioactivity	Electromagnetism
		formed when	produced it	mechanisms,	Use the idea that	How to use	non-iuminous objects	Acceleration	Atomic models	Liectionagnetism
		the light from a	Find patterns	including levers,	light travels in	switches in circuits	Describe how we see	Acceleration	Atomic models	Magnetic forces
		light source is	between the	pulleys and gears,	straight lines to	Switches in circuits	things	Velocity time graphs	Inside atoms	Wagnetie Torces
		blocked by an	volume of a	allow a smaller	explain why	Describe some	times			Electromagnetic Induction
		opaque object	sound and the	force to have a	shadows have the	models for circuits	Understand reflection,	Motion and Forces	Electrons and orbits	Electromagnetic induction
		opuque object	strength of the	greater effect.	same shape as the	and their limitations	refraction and	inotion and rorees		
		Find patterns in	vibrations that	Sicular circle.	objects that cast		dispersion	Resultant forces	Background radiation	The national grid
		the way that the	produced it	Earth and Space:	them.	Measure current				
		size of shadows		Describe the		and voltage with	Understand how we	Newtons first law	Types of radiation	Transformers and energy
		change.	Recognise that	movement of the	Electricity:	ammeters and	see colour			
			sounds get	Earth, and other	Associate the	voltmeters		Newtons second law	Radioactive decay	Particle Model
		Forces and	fainter as the	planets, relative	brightness of a		Energy transfers			
		Magnets:	distance from	to the Sun in the	lamp or the	Describe dangers of		Newtons third law	Half life	Particles and density
		Compare how	the sound source	solar system	volume of a buzzer	electricity and ways	Understand how to			
		things move on	increases.		with the number	to stay safe	measure temperature	Mass and weight	Using radioactivity	Energy and changes of state
		different		Describe the	and voltage of		changes			
		surfaces	Electricity:	movement of the	cells used in the	<u>Forces</u>		Momentum	Dangers of radioactivity	Energy calculations
			Identify common	Moon relative to	circuit		Describe conduction,			
		Notice that	appliances that	the Earth	Compare and give	Give examples of	convection and	Stopping distances	Radioactivity in medicine	Gas temperature and
		some forces	run on electricity		reasons for	forces	radiation			pressure
		need contact	Constructor	Describe the Sun,	variations in how		Describe in substitut	Braking distance and	Nuclear energy	Commence and a strength
		between two	Construct a	Earth and Moon	components	Use a newtonmeter	Describe insulation	energy	Nuclear finite	Gas pressure and volume
		objects, but	simple series	as approximately	function, including	to measure forces	Understand how to	Crash hazards	Nuclear fission	Foress and Matter
		magnetic forces can act at a	electrical circuit, identifying and	spherical bodies	the brightness of bulbs, the	Understand what	calculate efficiency and	Clashingzalus	Nuclear fusion	Forces and Matter Bending and stretching
		distance	naming its basic	Use the idea of	loudness of	friction does	how payments for	Waves	Nuclear Jusion	
		uistance	parts, including	the Earth's	buzzers and the	inction does	energy are calculated	Describing waves	Astronomy	Extension and energy
		Observe how	cells, wires,	rotation to	on/off position of	Understand how			<u></u>	transfers
		magnets attract	bulbs, switches	explain day and	switches	pressure is made	Earth and space	Wave speed	The solar system	
		or repel each	and buzzers	night and the	Use recognised	and the equation				Pressure in fluids
		other and	Identify whether	apparent	symbols when	for pressure	Evidence for the	Refraction	Gravity and orbits	Pressure and upthrust
		attract some	or not a lamp	movement of the	representing a		heliocentric model of			
		materials and	will light in a	sun across the	simple circuit in a	Understand	the solar system	Waves crossing	Lifecycle of stars	
		not others	simple series	sky.	diagram.	balanced and		boundaries		
			circuit, based on			unbalanced forces	Describe how seasons		Red shift	
		Compare and	whether or not				happen	Ears and hearing		
		group together a	the lamp is part			<u>Sound</u>			Origins of the universe	
		variety of	of a complete				How the Earth has a	Ultrasound		
		everyday	loop with a			Understand how	magnetic field		Energy – Forces Doing	
		materials on the	battery			sounds are made		Infrasound	Work	
		basis of whether	December that			Describe karry	How gravity varies		Work and power	
		they are	Recognise that a			Describe how	Dowood the selen		Forego and their Effects	
		attracted to a magnet, and	switch opens and closes a			soundwaves can be	Beyond the solar		Forces and their Effects	
		identify some	circuit and			seen	system		Objects affecting each	
		magnetic	associate this			Identify the			other	
		materials	with whether or			structure of the ear				
	l.	materials							<u> </u>	1

Describe magnets as having two poles Predict whether two magnets	not a lamp lights in a simple series circuit Recognise some common conductors and	Describe how sound can be used by animals and humans		Vector diagrams Rotational forces <u>Static Electricity</u> Charges and static electricity	
repel each other,	insulators, and associate metals with being good conductors.				